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Modern Chemistry 74 Quiz Section Quiz: The Gas Laws In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. ... Modern Chemistry 75 Quiz Name Class Date Section Quiz, continued ____ 6. The equivalent value of 273.15 K is a. 273.15 C. b. 273.15 C. c. 0 C. d. 25 C.

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Quiz Modern Chemistry 33 . Name Section Quiz, continued Class ... formed from ions in the gas state is known as a. thermal energy. b. lattice energy. c. ionization energy. d. radiant energy. 9. Some ionic compounds do not dissolve in water because ... Section Quiz: Metallic Bonding

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GASES Section Quiz: The Kinetic-Molecular Theory of Matter In the space provided, write the letter of the term or phrase that best completes each statement or answers each question. 1. A hypothetical gas that perfectly fits all the assumptions of the kinetic- molecular theory is known as a. a real gas. b. an ideal gas. c. an imaginary gas.

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Modern Chemistry 76 Quiz Section Quiz: Gas Volumes and the Ideal Gas Law In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. ____ 1.

Modern Chemistry Section 4 Solids Quiz - Maharashtra

Modern Chemistry Chapter 11 GASES. Section 1. Gases & Pressure. ... states that the total pressure exerted by a gas mixture is the sum of the partial pressures of the component gases. ... Read the Chemistry in Action section on page 368 of the textbook. Answer questions #1 & #2 at the end of the Chemistry in Action section on page 368. The Gas ...

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Modern Chemistry 76 Quiz Section Quiz: Gas Volumes and the Ideal Gas Law In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. ____ 1. At the same temperature and pressure, balloons of equal volume always contain a. equal masses of gas. b. equal numbers of molecules. c.

Modern Chemistry - Chapter - Gases - Section Quiz - Gas ...

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Modern Chemistry 72 Quiz Section Quiz: Gases and Pressure In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. ____ 1. What causes a gas to exert pressure? a. collisions b. density c. temperature d. elevation ____ 2. The SI unit for pressure is a. newton.

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Modern Chemistry 74 Quiz Section Quiz: The Gas Laws In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. ____ 1. For a fixed amount of gas at a constant temperature, the volume increases as the pressure a. remains steady.

Modern Chemistry Gases Section Quiz

Over four hundred years, scientists including Rudolf Clausius and James Clerk Maxwell developed the kinetic-molecular theory (KMT) of gases, which describes how molecule properties relate to the macroscopic behaviors of an ideal gas—a theoretical gas that always obeys the ideal gas equation. KMT provides assumptions about molecule behavior that can be used both as the basis for other ...

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